Lesson Plan

Name o	of the Institute:	C. V. Raman Polytechnic					
Department:		Basic Science					
Semester/Division/Branch:		1st sem/All Branches	r				
Subject Name with code: Total No. of Class (Required): Faculty Name:		Engineering Chemistry (Th-2b) 60 Dr Jayashree Samantray					
				Class N		the Topic/Chapter to be taught	Remarks
				1	Chapter 1: Atomic structure	: Fundamental particles	
2	Rutherford's Atomic model						
3	Atomic mass and mass number	r					
4	Bohr's Atomic model						
5	Bohr-Bury scheme, Aufbau's p	orinciple, Hund's rule					
6	Electronic configuration (up to	atomic no 30)					
7	Chemical Bonding : Definition	n , types					
8	Electrovalent bond with example	les					
9	Covalent bond with examples	19 19 19 19 19 19 19 19 19 19 19 19 19 19 1					
10	Coordinate bond with examples						
11	Acid base theory: Concept of	Arrhenius theory for acid and base with					
12	Lowry Bronsted theory for acid	and base with examples					
13	Lewis theory for acid and base w	vith examples					
14	Definition of Salt, Types of salts	,					
15	Solutions: Definitions of atomic	c weight, molecular weight, Equivalent					
16	weight Determination of equivalent weight	ght of Acid, Base and Salt.					
17	Modes of expression of the con	ncentrations (Molarity , Normality &					
18	Molality) pH of solution & Importance of p	H in industry					
19	Electrochemistry: Definition and	d types					
20	Faraday's 1st and 2nd law of Electronic	rolysis					
21	Corrosion: Definition of Corrosion	On. Types of Corresion	-				

22	Mechanism of rusting of Iron only. Protection from Corrosion	
23	Metallurgy: Definition of Mineral, ores	and the second s
24	Distinction between Ores And Minerals	
25	General methods of extraction of metals	
26	Concentration (Gravity separation, magnetic separation, Froth floatation & leaching)	
27	Oxidation (Calcinations, Roasting)	
28	Reduction (Smelting, Definition & examples of flux, slag)	
29	Refining of the metal (Electro refining, & Distillation only)	
30	Alloys: Definition of alloy. Types of alloys	
31	Hydrocarbons: Saturated and Unsaturated Hydrocarbons	
32	Aliphatic and Aromatic Hydrocarbons	
33	IUPAC system of nomenclature of Alkane, Alkene, Alkyne	
34	IUPAC system of nomenclature of alkyl halide and alcohol	
35	Bond line notation	
36	Uses of some common aromatic compounds (Benzene, Toluene, BHC)	
37	Uses of some common aromatic compounds (B Phenol, Naphthalene, Anthracene and Benzoic acid)	
38	Doubt Clearing Class	
39	Doubt Clearing Class	
40	Doubt Clearing Class	
41	Water Treatment : Sources of water	
42	Soft water, Hard water, hardness, types of Hardness (temporary or carbonate and permanent or non-carbonate)	
43	Removal of hardness by lime soda method (hot lime & cold lime— Principle, process & advantages)	
44	Advantages of Hot lime over cold lime process	
45	Organic Ion exchange method (principle, process, and regeneration of exhausted resins)	,
46	Lubricants: Definition of lubricant	
47	Types and specific uses of lubricants	
48	Purpose of lubrication	
49	Fuel: Definition and classification of fuel	
50	Liquid: Diesel, Petrol, and Kerosene Composition and uses	-
51	Gaseous: Producer gas and Water gas	

52	Elementary idea about LPG, CNG and coal gas (Composition and uses only)	
53	Polymer: Definition of Monomer, Polymer, Homo-polymer, Co-polymer and Degree of polymerization	
54	Uses of Polythene & Poly-Vinyl Chloride and Bakelite.	and the second s
55	Definition of Elastomer (Rubber). Natural Rubber (its drawbacks).	
56	Vulcanisation of Rubber. Advantages of Vulcanised rubber over raw rubber.	
57	Revision	and the second second
58	Revision	
59	Doubt Clearing Class	
60	Doubt Clearing Class	

Signature of the Faculty

Signature of the H.O.D