Lesson Plan

Name of the Institute:		C. V. Raman Polytechnic, BBSR			
Department:		Basic Science			
Semester	/Division/Branch:	1st sem/All Branches			
Subject I	Name with code:	Applied Chemistry			
Total No	. of Class (Required):	60			
Faculty N	Name:	Dr. Jayashree Samantray			
Class No.	. Brief description of	the Topic/Chapter to be taught	Remarks		
1	Unit 1: Atomic Structure, (
2	☐ Rutherford Model of an Atom				
3	□ Bohr's Theory				
4	☐ Types of Bonding				
5	☐ Electronic Configuration				
6		F hybridization in BeCl ₂ , BF ₃ , CH ₄ ,			
	NH ₃ , H ₂ O)				
7	☐ Coordinate bond				
8	☐ Hydrogen Bonding				
9	☐ Metallic Bonding				
10	☐ An Introduction☐ The idea of Solute, Solver	et and Solution			
11	1 0	oncentration of Solution			
12	Unit 2: Water- An Introduct				
13	Unit 2: Water- All introduct	Water Distribution on Earth			
14	- 21 '2 ' CO - A - 1	Hard Water			
15					
16	☐ Salts Causing Water Hardness ☐ Unit of Hardness				
17	OXY 1XX				
18	☐ Causes of Hard Water ☐ Cause of Poor Lathering of Soap in Hard Water				
19	Cause of Poor Lathering of	Top of Hard Water in Boiler			
20	 □ Problems Caused by the Use of Hard Water in Boiler □ Quantitative Determination of Water Hardness by ETDA 				
21		If of water flataness by 21211			
20	Method ☐ Water Softening Technique	es – An Introduction			
22	- 0 1 T' D				
23					
24	☐ Zeolite Process☐ Ion Exchange Process for `	Water Softening			
25	To the trial Tours of the own	at- An Introduction			
26		t- / III IIII oddouo			
27	☐ Screening				
28	□ Sedimentation				
29	☐ Coagulation				
30	☐ Filtration				

31	☐ Disinfection / Sterilization		
32	☐ Indian Standard Specification of Drinking Water – An		
	Introduction		
33	□ Water for Human Consumption		
34	Unit 3: Engineering Materials		
35	☐ Minerals and Ores		
36	☐ General Principles of Metallurgy		
37	☐ Extraction of Iron from Haematite ore		
38	☐ Extraction of Aluminium from Bauxite		
39	Alloy- Chemical Composition, Composition Based Applications		
40	☐ Portland Cement, Glasses, Refractory, Composite Materials		
41	Polymers		
42	☐ Preparation of Thermoplastics and Thermosetting Plastics		
43	□ Vulcanization of Rubber		
44	Unit 4: Chemistry Of Fuels And Lubricants		
45	☐ Fuel and Combustion		
46	☐ Classification of Fuels		
47	☐ Calorific Values (HCV and LCV)		
48	☐ Calculation of HCV and LCV using Dulong's formula.		
49	☐ Analysis of Coal		
50	Provimate Analysis of Coal (Solid Fuel)		
51	— Fuel rating of Petrol and Diesel (Octane and Cetane Numbers)		
52	☐ Chemical Composition, Calorific Values and Applications of		
	Fuel		
53	Unit 5 : Electro Chemistry		
54	Electrolytes and Non Electrolytes		
55	Application of Redox Reactions in Electrochemical Cells		
56	Corrosion – An Introduction		
57	Factors influencing Rate of corrosion		
	Internal Corrosion Preventive Measures		
58	External Corrosion Preventive Measures		
59	Revision		

Tayashree Sancatoay
Signature of the Faculty

Signature of the H.O.D